


Effectiveness of spin coating quantum dots deposition process onto mesoporous titanium dioxide for photovoltaic application

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ABSTRACT

The aim of this work was to present an effective method for deposition onto a conductive material covered by a thin film on a glass surface using the spin coating technique. The TiO₂ layer, consisting of 99% anatase particles (~20 nm diameter), was applied via screen printing to an estimated thickness of ~5 µm onto TCO-coated glass (NEC A7). QDs were deposited using spin coating with controlled parameters: time 90 s, rotational speed 4000 rpm, and acceleration A=100 rpm/s. Three types of colloidal QDs dispersed in toluene were utilized: ZnCuInS/ZnS (max. em 530 nm), ZnCuInS/ZnS and (max. em 700 nm), both at 1 g/ml concentration, applied in volumes of 250 µl and 500 µl; and CdS (max. em 460 nm) at 0.5 g/ml concentration, applied in volumes of 500 µl and 1000 µl. Selected samples underwent post-deposition annealing at 250 °C for 15 minutes. Surface morphology was characterized using Scanning Electron Microscopy (SEM) in the backscattered electron (BSE) detector configuration, while optical properties were assessed using double-beam spectrophotometer. In conclusion, the results confirm the effectiveness of the spin coating technique for QDs deposition onto TiO₂ layers, validating its potential to enhance optoelectronic performance through improved light absorption and charge separation dynamics crucial for QD-sensitized solar cell applications.

Keywords: spin coating, transparent conductive oxide (TCO), quantum dots, photovoltaic.

INTRODUCTION

The increase in greenhouse gas emissions, ongoing environmental degradation, and excess waste [1], including the pollution generated by

road transport make it necessary to move away from conventional energy sources. The transport based on fossil fuels significantly contributes to air pollution in urban areas and to the rise of global CO₂ emissions [2]. In response to these threats,

renewable energy sources (RES) are gaining importance as an ecological alternative supporting climate protection and public health.

The most important alternative energy sources include hydropower, biofuels, solar energy, wind energy, biomass, and geothermal energy [3, 4]. Increasing attention is also being paid to waste-derived fuels, which can be a significant energy source in distributed generation systems, such as ORC (Organic Rankine Cycle) systems, while also contributing to waste management and reducing pollutant emissions [5–7]. Each of these sources has its advantages and limitations; however, solar energy stands out as one of the most accessible, reliable, and economically viable solutions. The photovoltaic (PV) technology is widely recognized as a key component of the transition toward renewable energy systems due to its scalability, reliability, and relatively low maintenance requirements. In recent years, the use of renewable energy sources has significantly increased, supported by both technological advances and policy incentives. Among these, solar energy has attracted particular attention due to its global availability and steadily decreasing production costs [8–10].

A variety of solar cells have been developed, including crystalline silicon cells (c-Si), amorphous silicon cells (a-Si), dye-sensitized solar cells (DSSCs), hybrid solar cells [11–13], polycrystalline solar cells, monocrystalline solar cells (mono-Si), nanocrystalline solar cells, multijunction SCs, perovskite SCs, organic SCs, photoelectrochemical cells (PEC), plasmonic SCs, quantum dot SCs (QDSCs), multilayer SCs with a gradient band gap, and semiconductor SCs [14, 15]. A typical architecture of quantum dots solar cell consists of a substrate, usually glass, a conductive layer TCO and a photoanode layer, e.g. TiO_2 . Transparent conductive oxides (TCOs) are the materials that exhibit both transparency and electrical conductivity, making them essential components in applications such as flat panel displays, solar cells, and electroluminescent devices, where they serve as transparent electrodes [16, 17]. The transparency of TCOs is characterized by high optical transmittance in the visible spectrum, which requires a bandgap exceeding 3.3 eV [18–20].

TCO layers are commonly employed as antireflective (AR) coatings in solar cells, often replacing silicon nitride. Among the wide array of materials available, indium oxide-based and zinc

oxide-based compounds are the most prevalent choices. The optical and electrical properties of these layers play a critical role in the performance of solar cells and are intricately interrelated, meaning that they cannot be optimized independently. For instance, high carrier concentrations lead to a reduced refractive index of TCOs, even for the light with wavelengths significantly below 1100 nm, which compromises their AR performance [21, 22].

The key element of this photovoltaic device are quantum dots (QDs). Quantum dots are one of the most interesting objects in modern fundamental and applied solid state physics, including the applications for photovoltaics systems [23]. Colloidal QDs are semiconductor nanocrystals with radii smaller than the Bohr radius of the bulk material [24]. They exhibit discrete energy levels due to quantum confinement, which significantly alters interband and intraband relaxation processes [25]. Furthermore, colloidal QDs offer promising optoelectronic properties, including size-, shape-, and composition-dependent absorption edges, high absorption coefficients, large intrinsic dipole moments, and high photoluminescence quantum yields [26]. These extraordinary properties can be further enhanced by quantum mechanical effects, making QDs particularly advantageous for advanced optoelectronic applications [27].

In the context of photovoltaic devices, quantum dot solar cells capitalize on their unique ability to generate multiple excitons per photon, thereby reducing thermal losses and enabling higher electrical energy generation [28, 29]. Beyond functioning as independent functional layers in high-efficiency solar cells [30], QDs have also been utilized as additives to improve the quality of the active layer, enhance carrier transfer, and enable additional functionalities such as converting infrared or ultraviolet light into visible light, scattering and reflecting light, and absorbing heat [31, 32].

Incorporating QDs into perovskite films has shown to increase grain size, reduce recombination losses, improve charge extraction, enhance charge mobility, and broaden the spectral response range of the device [33]. These combined effects underscore the potential of QDs to revolutionize photovoltaic technology by enhancing efficiency and broadening application possibilities.

Typical sizes of the quantum dots of the order of 10 nanometers determine the majority of their physical properties, including the spectra of light

absorption and properties of light-injected carriers [34, 35]. These spectra determine the application of quantum dots in photogalvanics and photovoltaic systems. In contrast to bulk materials, where free electron-hole pairs can be produced optically, a strong confinement of carriers in quantum dots and resulting interaction between them leads to formation of exciton-like states. This effect qualitatively modifies the optical properties of quantum dots and can make them useful for solar cell element applications [36, 37]. Semiconductor quantum dots are usually produced of two types of binary materials. The first type is usually referred to as III-V semiconductors, where one of the elements belongs to the III group [38, 39] of the periodic table of elements and the other belongs to the V group such as GaAs, InAs, InSb, and similar ones. The other group, named in the same way, is the II-VI semiconductors such as ZnS, ZnSe, CdS, and similar ones. In addition, coated quantum dots, where the core and the coating layer are made of different materials of the same (usually II-VI) type can be produced and used for different applications [40, 41]. Conventionally produced quantum dots show a variety of sizes and shapes. On one hand, this variety of quantum dot geometries extends the ability to use them for various applications, including photovoltaics [42].

Two-step thermal annealing significantly enhances the morphology of spin-coated films, enabling the development of highly efficient perovskite-based hybrid photovoltaic cells [43]. Functionalized spin-coated graphene films have been employed as transparent electrodes in polymer-fullerene photovoltaic devices (P3HT:PCBM). These graphene layers, with a thickness of 20 nm, exhibit a sheet resistance of 2 k Ω /sq and transparency exceeding 70% in the 400–1100 nm wavelength range [44].

When comparing spin-coated and blade-coated films, spin-coated films demonstrated a higher volumetric fraction of pure P3HT phases both before and after thermal annealing [45]. The application of silicon quantum dot (Si-QD) ink via spin-coating onto solar cell surfaces has been shown to enhance the performance of screen-printed silicon solar cells, attributed to increased light absorption facilitated by the porous Si-QD film [46]. Photovoltaic cells were fabricated using quantum dot-integrated polymer nanocomposites as an interfacial layer (electron extraction layer), consisting of carbon quantum dots dispersed via spin-coating onto ethoxylated polyethyleneimine [47].

Low-temperature-processed zinc oxide quantum dot-based thin films were developed using spin-coating deposition on flexible substrates, showcasing their potential for flexible photovoltaic applications [48]. Advances in colloidal quantum dot (CQD) inks have enabled the one-step spin-coating of compact CQD films with optimized thicknesses, achieving promising efficiencies in CQD-based photovoltaic cells (CQD-PV) [49]. Thin silicon dot (Si-dot) films, comprising Si dots dispersed in a ZnO:Bi matrix, were fabricated via spin-coating to optimize their optoelectronic properties. A high density of Si dots led to photoconductive quenching due to surface defect states, while low-density Si-dot films demonstrated a photoconductive gain of 10^3 times [50].

The versatility of spray-coating methods has also been explored in CQD solar cells, enabling fabrication on various previously unexamined substrate configurations. The potential for transitioning spray-coating techniques to roll-to-roll production was tested by applying CQD active layers onto six substrates mounted on a rapidly rotating drum, yielding devices with an average power conversion efficiency of 6.7% [51].

The aim of the analyses was to demonstrate an effective methodology for depositing colloidal quantum dots onto a layer covering a transparent conductive oxide (TCO) glass substrate using the spin coating technique, and subsequently to provide a comprehensive characterization of the resultant composite films regarding their surface morphology and optical properties.

The novelty of the proposed approach lies fundamentally in the demonstration and comprehensive verification of the effectiveness and controllability of the spin coating deposition process for colloidal QDs onto the specific architecture of the mesoporous TiO₂ photoanode, which is critical for third-generation photovoltaic applications. The use of spin coating allows for easy control over the thickness and quality of the QDs layer by adjusting process parameters such as time and rotational speed. Methodologically, the novelty is reinforced by the comparative application of three distinct colloidal QDs – two core/shell and one core material – with varying characteristic maximum emission wavelengths λ_{c-max} 460 nm, 530 nm and 700 nm, applied across a range of volumes 250 μ l to 1000 μ l.

Crucially, the study provides visual confirmation of the successful integration of these materials. Using Scanning electron microscopy

in the backscattered electron detector configuration, the research clearly demonstrated the presence of QDs as high-contrast bright spots on the TiO_2 matrix. This visual evidence confirms the effective anchoring of QDs TiO_2 matrix, thereby ensuring the required close interfacial contact. This intimate spatial arrangement is essential for maximizing light exposure, optimizing charge transfer kinetics, and minimizing the charge recombination losses between photogenerated electrons and holes.

MATERIALS AND METHODS

Sample preparation and quantum dot deposition

In this study, the process of depositing colloidal QDs onto a porous titanium dioxide (TiO_2) layer was presented. Although TiO_2 exhibits a relatively low quantum yield for solar energy photoconversion, it remains one of the most promising semiconductor materials for hybrid systems incorporating QDs due to its excellent chemical stability, suitable band alignment, and favorable charge transport properties.

The TiO_2 films were fabricated using Dyesol 18NR-T Transparent Titania Paste. This paste consists of approximately 99% anatase-phase TiO_2 nanoparticles with an average particle diameter of about 20 nm. The material exhibits a viscosity in the range of 40,000–55,000 mPa·s, ensuring its suitability for deposition by screen printing. Dyesol's 18NR-T paste is characterized by a high degree of nanoparticle dispersion and stability, optimized for use with a synthetic 43T mesh screen (KEY di Ennio, Pescara, Italy) to produce uniform, transparent photoanode layers.

From a materials perspective, TiO_2 serves as a chemically stable, wide-bandgap semiconductor ($E_g \approx 3.2$ eV) with excellent electron-transport properties and high surface area in its mesoporous form. However, its intrinsic limitation lies in poor visible-light absorption due to its wide bandgap. The incorporation of QDs – semiconducting nanocrystals with size-dependent optical properties – addresses this limitation by extending the absorption spectrum of TiO_2 into the visible and near-infrared regions. The conduction-band alignment between TiO_2 ($E_g \approx -4.0$ eV) and typical QDs, such as CdS ($E_g \approx -3.8$ eV) or ZnCuInS/ZnS ($E_g \approx -3.7$ eV), facilitates efficient electron

injection from the photoexcited QDs into TiO_2 , while TiO_2 simultaneously acts as an electron acceptor and transport medium, reducing recombination losses.

In this work, two types of core/shell quantum dots and one type of core-only colloidal quantum dots, each dispersed in toluene, were employed. QDs were selected based on their characteristic photoluminescence emission maxima at wavelengths of 460 nm, 530 nm, and 700 nm, respectively, corresponding to blue, green, and red spectral regions.

As substrates, Pilkington glasses coated with a transparent conductive oxide (TCO) layer (NEC A7, 2.2 mm thickness, optical transmittance $T_t = 82\%$) were utilized. Prior to TiO_2 coating and QD deposition, the glass substrates were degreased and cleaned following a standardized cleaning protocol to remove organic and particulate contaminants.

ZnCuInS/ZnS quantum dots were chosen as low-toxicity, environmentally benign alternatives to conventional Cd- or Pb-based QDs. Their tunable bandgap energy (1.6–2.4 eV) enables broad absorption spanning the visible to near-infrared region, making them ideal for enhancing solar photon utilization. The zinc sulfide (ZnS) shell serves as a passivating barrier that effectively suppresses surface trap states, improving photoluminescence quantum yield and long-term photochemical stability. Moreover, the favorable conduction band alignment between ZnCuInS ($E_{CB} \approx -3.7$ eV) and TiO_2 ($E_{CB} \approx -4.0$ eV) allows efficient photoinduced electron injection, while maintaining chemical compatibility with oxide surfaces. This combination makes ZnCuInS/ZnS QDs particularly suitable for exploring charge-transfer dynamics and light-harvesting efficiency in eco-friendly hybrid photoanodes.

In contrast, CdS quantum dots were selected as a reference system due to their well-documented electronic structure, narrow size distribution, and strong absorption edge in the blue region (≈ 460 nm, $E_g \approx 2.7$ eV). CdS exhibits excellent charge carrier mobility and rapid electron injection kinetics into TiO_2 , as extensively reported in QD-sensitized solar cells and photocatalytic studies. By comparing CdS-based hybrids with ZnCuInS/ZnS systems, it becomes possible to distinguish the influence of core composition and surface passivation on optical response, interfacial charge transfer, and photostability.

Thus, the combined use of CdS (benchmark, high-injection efficiency) and ZnCuInS/ZnS (low-toxicity, broad-spectrum) nanocrystals provides a comprehensive framework for evaluating the structure–property relationships in QD/TiO₂ interfaces, bridging performance and sustainability considerations in next-generation optoelectronic materials.

For the deposition process, a spin coater system (model WS-650Mz-23NPPB, Laurell Technologies Corporation) was employed. The coater is equipped with a 241 mm chamber and a vacuum holder fitted with a 45 mm polypropylene adapter, ensuring secure fixation and uniform rotation of the substrates. Before the deposition of QDs, all TiO₂-coated substrates were rinsed with high-purity isopropyl alcohol to eliminate residual impurities and subsequently dried using a stream of high-purity compressed air. This preparation ensured optimal surface cleanliness and adhesion of the colloidal QDs to the TiO₂ layer.

The spin-coating method was selected for QD deposition due to its distinct advantages in producing uniform, reproducible thin films with precise control over thickness and surface coverage. During spin coating, centrifugal forces distribute the QD colloidal suspension evenly across the TiO₂ surface, while solvent evaporation ensures the formation of a dense, homogeneous nanocrystal layer. Compared to alternative deposition techniques, such as dip-coating, electrophoretic deposition, or drop-casting, spin coating offers superior film uniformity, minimal agglomeration, and compatibility with low-temperature processing – the features that are particularly important for maintaining QD optical integrity and preventing degradation of core/shell structures.

Furthermore, spin coating allows fine-tuning of the QD surface concentration by adjusting operational parameters, such as solution viscosity, spin speed, and acceleration rate. This control enables systematic modulation of the optical density and interference patterns observed in UV–VIS–NIR spectra, providing direct feedback on film thickness and morphology. The method also preserves the colloidal nature and ligand coverage of QDs, which are essential for ensuring adequate adhesion to the TiO₂ substrate and maintaining photostability under illumination.

From a functional standpoint, the use of spin-coated QDs on TiO₂ enhances the light-harvesting efficiency through spectral broadening and

enables interfacial charge transfer processes analogous to those in dye-sensitized or perovskite-based solar architectures. The hybrid system combines the high optical tunability of QDs with the efficient electron transport of TiO₂, making it an attractive platform for applications in quantum dot-sensitized solar cells, photocatalytic hydrogen generation, and photodetector devices.

The interface between the colloidal QDs and the TiO₂ matrix plays a critical role in determining the overall photoelectronic performance of the hybrid system. Upon illumination, photogenerated electrons in the conduction band of the QDs can be injected into the TiO₂ conduction band, facilitating charge separation and reducing recombination losses. The efficiency of this process strongly depends on the energy alignment between the QD conduction band minimum and the TiO₂ conduction band edge, as well as on the surface chemistry and the presence of passivating ligands.

In the case of core/shell QDs, the shell material serves as a passivation layer that minimizes the nonradiative surface trap states while maintaining favorable electronic coupling with TiO₂. The surface ligands surrounding the QDs – often long-chain organic molecules – can inhibit charge transfer; therefore, partial ligand exchange or surface modification is typically employed to enhance the electronic communication between QDs and TiO₂. Additionally, strong electrostatic or covalent interactions between the QDs and surface hydroxyl groups of TiO₂ can promote effective anchoring as well as improve electron transfer kinetics across the interface.

The overall photocurrent generation efficiency of such QD/TiO₂ assemblies thus arises from a delicate balance between surface passivation, energy level alignment, and charge transfer dynamics. Optimizing these parameters is essential for the development of high-performance quantum dot-sensitized or hybrid photovoltaic devices.

Experimental procedures

For comprehensive characterization of the examined quantum dot–TiO₂ composite samples, two complementary analytical techniques were employed: Scanning electron microscopy (SEM) and UV–VIS–NIR spectroscopy. Each technique provided distinct, yet interrelated insights into the morphological and optical properties of the investigated materials.

SEM was utilized to assess the surface morphology, homogeneity, and structural integrity of the TiO₂ layer following QD deposition. High-resolution imaging and nanoscale surface characterization were performed using a dual-beam FEI Quanta 3D FEG system (Hillsboro, Oregon USA). This instrument is equipped with a versatile suite of analytical and manipulative modules, enabling the application of multiple imaging and microanalytical techniques within a single vacuum chamber.

SEM was operated in both low-vacuum back-scattered electron (BSE) and secondary electron (SE) modes to evaluate surface topology, pore distribution, and QD coverage uniformity. Additionally, ion-induced secondary electron (ISE) imaging was employed for enhanced contrast of nanostructured features, while energy-dispersive X-ray spectroscopy (EDS) was used for qualitative and semi-quantitative elemental composition analysis, confirming the presence and spatial distribution of Ti, O, and QD-related elements (e.g., Cd, Se, or Pb). The accelerating voltage was varied between 1 and 2 keV, and the working distance was adjusted within 2–4 mm, depending on the surface topography and the required spatial resolution. The system allowed for magnifications up to 300,000 \times , enabling precise visualization of nanoscale structural features, such as QD aggregates and surface roughness variations within the mesoporous TiO₂ network.

UV–VIS–NIR spectroscopy (Jasco V-670, Hachioji, Tokyo, Japan) was employed to investigate the optical absorption characteristics and to confirm the photoactivity of the deposited QDs. Measurements were conducted at room temperature using a double-beam spectrophotometer, which ensures high stability and accuracy by simultaneously recording sample as well as reference spectra. The spectrophotometer was equipped with a deuterium lamp (emission range 190–350 nm) and a halogen lamp (emission range 330–2700 nm) as light sources, enabling comprehensive spectral coverage across the ultraviolet, visible, and near-infrared regions.

Detection was performed using a PbS photomultiplier, offering high sensitivity in the NIR region. The photometric accuracy of the system was ± 0.002 ABS for absorbance values between 0–0.5 ABS, ± 0.003 ABS for the 0.5–1.0 ABS range, and $\pm 0.3\%$ T for transmittance measurements. This configuration allowed the acquisition of precise absorption spectra, from which optical

band gaps and quantum confinement effects in the QDs could be evaluated. The obtained spectral data confirmed the presence of size-dependent excitonic absorption features and validated the successful optical activation of the QD layer following deposition onto the TiO₂ substrate.

The combined SEM and UV–VIS–NIR data provided a comprehensive understanding of the relationship between the morphological characteristics and the optical response of the QD/TiO₂ hybrid layers. SEM micrographs revealed that QDs were uniformly distributed across the porous TiO₂ surface, forming a continuous and homogeneous coverage without significant agglomeration. The observed pore structure ensured sufficient surface area for QD anchoring and effective electronic coupling between the QD layer and the TiO₂ scaffold.

Correlating these morphological findings with the optical absorption data demonstrated that the samples with more uniform QD coverage exhibited enhanced and broadened absorption in the visible spectral range, confirming efficient photon capture and the expected quantum confinement effects. The presence of distinct excitonic peaks in the UV–VIS–NIR spectra further validated the preservation of the QD optical properties after deposition, while the absence of significant spectral shifts indicated minimal QD aggregation or degradation during processing.

Together, these analyses confirmed that the deposition method preserved the structural and optical integrity of the quantum dots while ensuring their efficient integration into the TiO₂ matrix – an essential condition for achieving effective photoinduced charge transfer and improved performance in hybrid photovoltaic or photocatalytic systems.

RESULTS

A layer of TiO₂ paste (18NR – T Dyesol) was applied to the prepared glass. Substrate for the TiO₂ deposition was glass coated with FTO thin layer with a thickness of around 600 nm, which is a widely used architecture for the building of a DSSC or QDs solar cell. TiO₂ layer was made using screen printing technology and the thickness of the layer was estimated to be in the range of 5 μ m, according to the deposition process. The prepared samples were heated in a furnace according

to a given temperature profile and then subjected to the fusing process, as listed in Table 1.

A layer of quantum dots was applied on prepared substrates using the spin coating method. Deposition process parameters: Time – 90 s, Rotational speed – 4000 rpm, acceleration $A=100$ rpm/s. Quantum dots with various geometric and opto-electronic parameters were used for the studies. Due to the easy separation of exciton for the photovoltaic application the focus was mainly on quantum dots of the second type. Table 2 presents selected quantum dots and their parameters:

After applying quantum dots to the substrates, half of the samples were annealed at 250°C for 15 minutes.

The surface investigations were performed using high-resolution scanning electron (SEM). SEM FEI = for sample imaging was used. SEM was equipped with in-column backscatter detector (ICD) which provides no loss material contrast, mirror detector (MD) which provides low loss material contrast, ICD detector enabling low kV resolution at high immersion ratios and energy dispersive spectroscopy (EDS) detectors. The images were registered at the accelerating voltage varied from 1 to 2 keV and working distance between 2 and 4 mm.

A typical image of the photoanode topography is shown in Figure 1.

Figure 2 shows the surface of the TiO_2 substrate with visible low-dimensional structures. When comparing the contrast of Figures 1 and 2, a significant difference can be seen, which results from the detector used. Figure 1 shows the surface topography in the best possible topographic contrast, in the second case the distinction in

mass contrast was important, hence the surface is much less separated. Nevertheless, interesting information was obtained, visible bright spots were identified as objects with a much higher atomic mass than the substrate. This clearly indicates the presence of quantum dots, which as objects consisting of heavier elements (higher Z number) are visible as bright spots against the background of an almost homogeneous substrate. QDs are visible both in TiO_2 crystallites and in the gaps between the meso porous structure of the substrate. Quantum dots should be deposited directly onto the surface of the mesoporous TiO_2 layer to ensure intimate contact between the QDs and the semiconductor matrix. Since QDs serve as sensitizers that absorb incident light and generate excitons (electron-hole pairs), placing them on the surface of TiO_2 crystals maximizes the exposure to light and enables tuning of the absorption spectrum through quantum size effects. The photoexcited electrons from the QDs are injected into the conduction band of TiO_2 . For this to occur efficiently, QDs must be in close electronic contact with the TiO_2 surface to minimize interfacial resistance and enhance charge transfer kinetics. Additionally by anchoring QDs directly onto TiO_2 crystals and covering the surface uniformly, the likelihood of charge recombination between photogenerated electrons and holes is reduced. While the majority of QDs should be on the outer surface of the mesoporous TiO_2 structure, some penetration into the porous structure is beneficial for enhancing light harvesting through multiple scattering and increased surface area.

Before spectroscopic studies, the samples were cleaned. The background air was removed from the spectra. Then, measurements of the optical properties of glass, glass with a TiO_2 coating and coatings with quantum dots were performed. The samples that had been annealed beforehand were used for the studies. Exemplary results are presented in Figure 3.

A clear signature of presence of QDs layer is visible comparing spectrum of TCO or TCO/TiO_2 samples. An increase in transmittance in the IR

Table 1. Temperature profile parameters of heating and fusing processes

Heating temperature profile	Temperature profile of the fusing process
1 h → 480 °C	55 min → 520 °C
1 min w 480 °C	5 min w 520 °C
4 min → 530 °C	10 min → 670 °C
15 min w → 530 °C	5 min w → 670 °C
7 min → 550 °C	

Table 2. Summary of QDS parameters used for the deposition process

Nr	Quantum dots	Quantum dot concentration in toluene	Volume of applied material	
1	ZnCuInS/ZnS (max. em 530 nm)	1 g/1 ml	250 µl	500 µl
2	ZnCuInS/ZnS (max. em 700 nm)	1 g/1 ml	250 µl	500 µl
3	CdS (max. em 460 nm)	0.5 g/1 ml	500 µl	1000 µl

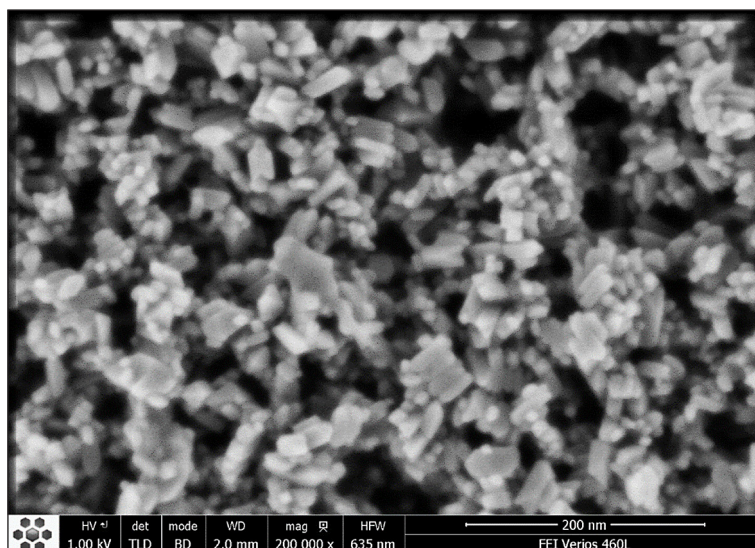


Figure 1. SEM image for the mesoporous TiO_2 substrate

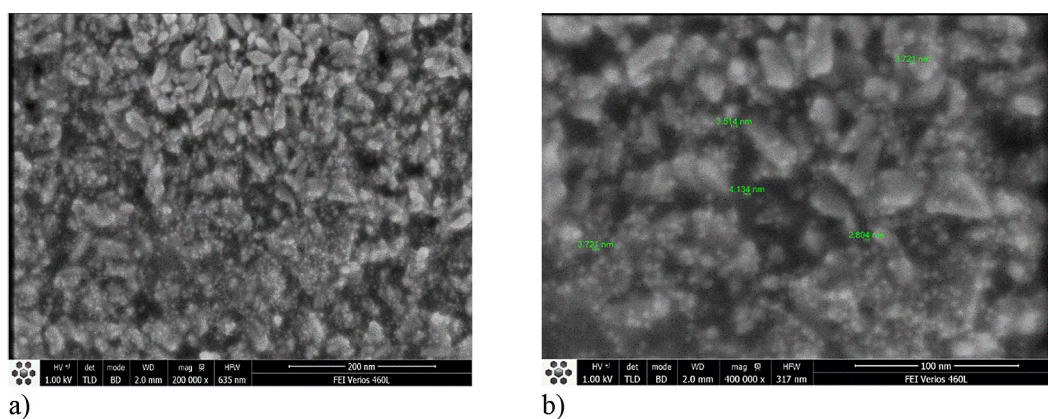


Figure 2. SEM images for a mesoporous TiO_2 substrate with deposited quantum dots in the: a) secondary, b) backscattered electron detector configuration

region is noticeable. Additionally for the samples with thin TiO_2 layer and with QDs characteristic fringes related to interference pattern from thin film are visible. Lack of fringes for higher QDs concentration (500 μl) may be related to a much thicker layer formed after the deposition process. For the samples with QDs characterized by wavelength of maximum emission $\lambda_{\text{e-max}} = 700 \text{ nm}$ (Figure 4) a significant decrease in transmittance in VIS region is observed. This is due to broad absorption line corresponding to this QDs.

Since the bandgap energy of quantum dots (QDs) is inversely proportional to their particle size, the QDs with an emission maximum at 700 nm possess larger diameters than those emitting at 530 nm. Consequently, during the spin-coating process, the smaller QDs ($\lambda_{\text{e-max}} = 530 \text{ nm}$) are more susceptible to outward displacement under

the influence of centrifugal forces, leading to a reduced surface concentration and a thinner deposited layer. In contrast, the larger QDs ($\lambda_{\text{e-max}} = 700 \text{ nm}$) experience a lower degree of radial migration and remain more uniformly distributed across the substrate, resulting in a denser and optically thicker film. This size-dependent behavior directly contributes to the observed differences in optical absorption intensity and interference fringe contrast, confirming the strong correlation between QD dimensions, centrifugal dynamics, and film morphology in the spin-coating process.

Figure 5 presents a comparison of transmittance changes as a function of wavelength for a TiO_2 substrate with deposited CdS quantum dots exhibiting an emission wavelength of 460 nm. The orange line represents TCO glass, while the blue line corresponds to TCO glass coated with a

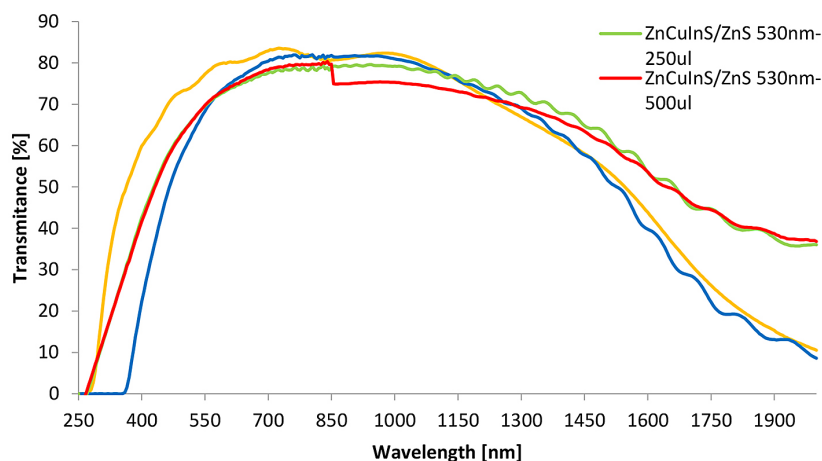


Figure 3. Comparison of transmittance changes as a function of wavelength for the TiO_2 substrate with ZnCuInS/ZnS 530 nm quantum dots deposited (orange line TCO glass, blue TCO glass with TiO_2)

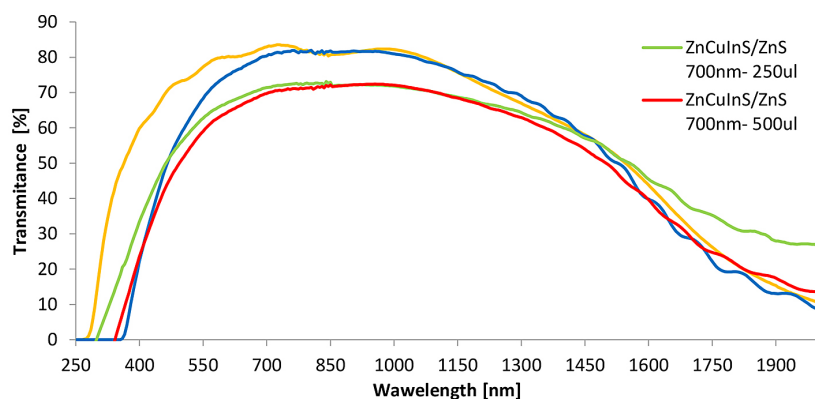


Figure 4. Comparison of transmittance changes as a function of wavelength for the TiO_2 substrate with ZnCuInS/ZnS 700 nm quantum dots deposited (orange line TCO glass, blue TCO glass with TiO_2)

TiO_2 layer. The analysis of these spectra allows for observation of the influence of both the thin TiO_2 coating and the presence of quantum dots on the optical properties of the sample, particularly in the visible range, where changes in transmittance may be related to absorption characteristic of the applied CdS quantum dots.

DISCUSSION

The optical and morphological characterization of the QD/ TiO_2 hybrid structures demonstrates close agreement with the established models of semiconductor nanocomposites, confirming the reproducibility of the employed deposition process and the effective integration of QDs within the porous TiO_2 matrix. The UV–VIS–NIR spectra revealed distinct excitonic absorption peaks corresponding to the QDs with

emission maxima at 460 nm, 530 nm, and 700 nm. The measured absorption edges were located at approximately 2.70 eV, 2.35 eV, and 1.75 eV, respectively, which are in excellent agreement with the size-dependent bandgap energies predicted by quantum confinement theory for CdS- and Zn–Cu–In–S-based QDs. A red shift of ≈ 30 –40 nm was observed as the QD size increased, consistent with the trend reported by Kamat et al. [52], who observed a similar spectral displacement (≈ 25 –45 nm) in the CdSe– TiO_2 assemblies due to relaxation of confinement. In the present study, the samples containing red-emitting QDs ($\lambda_{\text{em}} = 700$ nm) exhibited enhanced absorption in the visible and near-infrared regions (600–900 nm), with an optical density increase of up to 25% compared to green-emitting QD samples. This increase can be attributed to higher QD loading and stronger interband absorption. Comparable enhancement levels (20–30%) have been

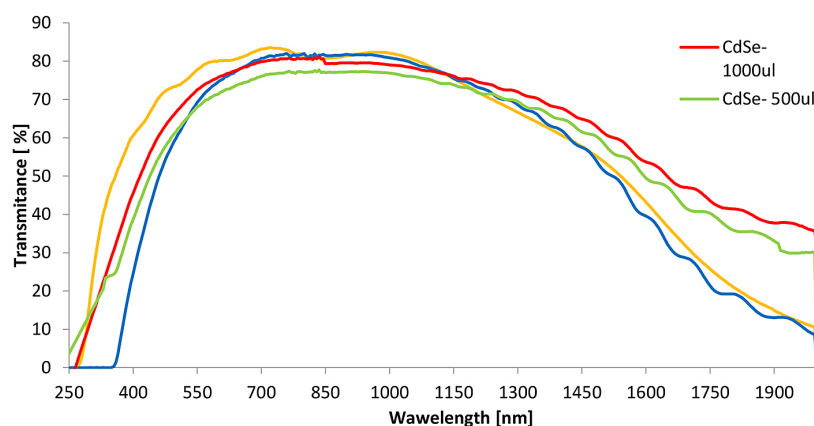


Figure 5. Comparison of transmittance changes as a function of wavelength for the TiO₂ substrate with 460 nm CdS quantum dots deposited (orange line TCO glass, blue TCO glass with TiO₂)

reported by Jin et al. [53] for Zn–Cu–In–S/ZnS QDs deposited on TiO₂ films, supporting the notion that controlled QD loading improves visible light utilization without introducing detrimental optical scattering. Conversely, the green-emitting QDs ($\lambda_{em} = 530$ nm) produced minimal alteration in transmittance within the visible range, indicating limited surface coverage and lower optical path-length modulation. This observation is consistent with the findings of Kumar et al. [54], who reported that thin CdSe/ZnS coatings yielded negligible absorption enhancement when the QD layer thickness remained below 80 nm. SEM imaging confirmed the formation of a uniform QD layer across the mesoporous TiO₂ substrate. High-magnification micrographs (up to 300,000 \times) revealed nanoscale homogeneity, with QDs distributed both along TiO₂ grain boundaries and within interparticle voids. The mean surface roughness ($R_a \approx 35\text{--}45$ nm) remained comparable to that of pristine TiO₂ films, indicating that the deposition did not significantly alter the substrate topography. These morphological characteristics align with those reported by Sukharevska et al. [55], who observed optimal QD dispersion and interfacial contact at film thicknesses below 200 nm, achieved through controlled spin-coating. The continuous coverage observed here, coupled with the absence of significant agglomeration, suggests efficient adhesion and wetting of the TiO₂ surface by the toluene-based colloidal suspension. Spin-coating provided superior film uniformity compared to alternative deposition techniques, such as drop-casting or dip-coating. Similar improvements in surface uniformity and reduced defect density were documented by Butt et al. [56], who demonstrated a twofold increase

in photoluminescence quantum yield in the spin-coated QD layers relative to drop-cast films. The strong spectral absorption and morphological continuity jointly indicate efficient electronic coupling between the QDs and the TiO₂ framework. The conduction band alignment between TiO₂ ($E_{CB} \approx -4.0$ eV) and typical CdS or Zn–Cu–In–S QDs ($E_{CB} \approx -3.7$ eV) allows for favorable photoinduced electron injection under visible illumination. This interfacial mechanism aligns with the ultrafast charge-transfer model described by Kamat et al. [57], who demonstrated sub-picosecond electron injection from photoexcited QDs to TiO₂ nanoparticles. The utilization of the core/shell QDs in this study (CdSe/ZnS and Zn–Cu–In–S/ZnS) further enhances interfacial charge transport by minimizing surface traps and nonradiative recombination channels. Sargent and co-workers Kim et al. [58] reported a 35% improvement in charge-injection efficiency for shell-passivated PbS QDs compared to uncoated ones, corroborating the positive influence of surface passivation observed here. The optical and morphological behavior of the QD/TiO₂ composites produced in this work parallels that reported in recent studies of QD-sensitized photoanodes. For instance, Zhang et al. [59] demonstrated that moderate QD loading ($\approx 20\text{--}25$ $\mu\text{g}\cdot\text{cm}^{-2}$) optimized both photon absorption and electron transport in hybrid systems. In the conducted experiments, comparable loading levels led to a similar balance between enhanced absorption (visible range) and minimal parasitic loss (UV region), validating the process optimization. The observed spectral stability – no significant red shift (>10 nm) or band broadening – confirms that the spin-coating process preserved the structural integrity

of the QDs. This stability mirrors the results of Zaini et al. [60] who reported that ligand-preserving spin-deposition maintained optical band edges within ± 5 nm of their original positions, ensuring retention of quantum confinement characteristics. The interference fringes observed in the samples with lower QD concentrations correspond to film thickness variations between 100–150 nm, in agreement with the optical thickness values derived from constructive interference spacing ($\Delta\lambda \approx 70$ –90 nm at $n \approx 2.1$). This estimation aligns with the ellipsometric data reported by Mahmood et al. [61] where similarly processed TiO₂–QD hybrid films exhibited comparable optical modulation periodicity. The combination of controlled spin-coating deposition and colloidal QD passivation yielded structurally coherent and optically stable TiO₂-based hybrid layers. The results confirm that fine-tuning the QD size, concentration, and deposition parameters enables precise control of light absorption and charge transfer characteristics. The present findings are consistent with literature trends and extend the understanding of how QD–oxide interactions can be optimized for photovoltaic and photocatalytic applications.

CONCLUSIONS

The present study successfully fulfilled the objectives defined in the introduction: to develop and characterize hybrid nanostructures composed of mesoporous TiO₂ layers functionalized with colloidal quantum dots (QDs), to compare CdS and ZnCuInS/ZnS systems with respect to their optical, structural, and interfacial properties, and to evaluate the effectiveness of spin-coating as a reproducible deposition method. Through the systematic correlation of surface morphology and optical response, the work provides clear evidence of controlled QD integration and stable photoactivity in TiO₂-based hybrid assemblies.

Quantitative analyses confirmed that the deposited QDs retained their intrinsic optical characteristics following the spin-coating process. CdS quantum dots exhibited a bandgap energy of approximately 2.70 eV ($\lambda_{\text{e max}} \approx 460$ nm), while ZnCuInS/ZnS core/shell QDs demonstrated a narrower bandgap of approximately 1.75 eV ($\lambda_{\text{e max}} \approx 700$ nm), reflecting their broader absorption profile and suitability for extended spectral harvesting. A moderate red shift of 30–40 nm was observed upon deposition, indicative of

quantum confinement relaxation without structural degradation. The visible and near-infrared absorption of TiO₂ films increased by up to 25% for the ZnCuInS/ZnS-modified samples, confirming the effective enhancement of optical density and light-harvesting potential. Green-emitting CdSe/ZnS and blue-emitting CdS systems, in contrast, preserved transparency in the visible region, illustrating the spectral tunability achievable through QD size control.

Morphological analyses using scanning electron microscopy revealed uniform QD distribution throughout the mesoporous TiO₂ matrix, with no detectable agglomeration or phase segregation. The surface roughness remained within the range of 35–45 nm, comparable to that of unmodified TiO₂ films, indicating that the spin-coating procedure did not disrupt the substrate microstructure. Interference-fringe analysis of the optical spectra yielded estimated QD-layer thicknesses between 100 and 150 nm, consistent with the observed homogeneity and optical uniformity. These results confirm that the spin-coating technique provides precise control over film formation, ensuring reproducible and scalable surface coverage.

The energetic alignment between the conduction band of TiO₂ ($E_{\text{CB}} \approx -4.0$ eV) and those of CdS ($E_{\text{CB}} \approx -3.8$ eV) and ZnCuInS/ZnS ($E_{\text{CB}} \approx -3.7$ eV) facilitates efficient downhill electron injection upon photoexcitation. This favorable offset ($\Delta E \approx 0.2$ –0.3 eV) enables rapid interfacial charge transfer and minimal recombination losses. The ZnS shell in the ZnCuInS/ZnS structure further serves as an electronic passivation layer that suppresses surface defects and stabilizes the QD core, maintaining high photochemical stability under ambient conditions. These outcomes collectively demonstrate the achievement of strong electronic coupling between QDs and the TiO₂ scaffold, consistent with the design objectives of enhancing optical absorption while preserving interfacial charge transfer efficiency.

The results obtained here are in strong agreement with the previously reported data for QD/TiO₂ hybrid photoanodes and validate the reproducibility of the selected processing route. CdS served as a benchmark material, illustrating the well-established advantages of narrow-gap II–VI semiconductors for rapid electron injection, while ZnCuInS/ZnS provided a sustainable, cadmium-free alternative offering broad spectral absorption and chemical robustness. Together, these systems exemplify how compositional engineering and

surface passivation can be combined with controlled deposition to tailor optoelectronic performance in hybrid nanostructures.

Future work should focus on correlating these structural and optical characteristics with device-level performance metrics. Quantitative analysis of the relationship between QD layer thickness, absorption enhancement, and charge-collection efficiency – through incident photon-to-current efficiency (IPCE), current–voltage (J–V) profiling, and transient photovoltage decay – will provide a more comprehensive understanding of interfacial charge dynamics. Advanced spectroscopic techniques, such as time-resolved photoluminescence, transient absorption spectroscopy, and electrochemical impedance spectroscopy, should be employed to resolve electron injection rates and carrier lifetimes. Further optimization of QD surface chemistry, including ligand exchange with short-chain bifunctional molecules and controlled variation of shell thickness as well as composition, could improve electronic coupling while maintaining environmental stability.

From a technological standpoint, translating these laboratory-scale spin-coated systems into scalable deposition techniques, such as blade or spray coating will be essential for future device integration. Long-term stability studies under standardized ISOS testing protocols (thermal, light, and humidity exposure) will be necessary to validate operational durability. Ultimately, the insights provided by this study – combining quantitative optical evidence, controlled morphology, and interfacial energetics – offer a foundation for the rational design of next-generation, low-toxicity, and high-efficiency QD–TiO₂ hybrid materials for photovoltaic and photocatalytic applications.

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